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4th EU + OECD Webinar on Hydrogen Fuel Risks Hydrogen Fuel Risks Associated with Ammonia as a Hydrogen Fuel Carrier March 11th, 2025



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- Ammonia in the Energy Transition
- Production, Use, and Future Outlook
- Safety Challenges and Risk Profiles
- Infrastructure, Operations, and Human Factors
- Conclusive remarks and Path Forward

Global production of ammonia (NH₃): **150 Mt** (2023). Regions with the highest production: East Asia (65 Mtons), Russia, US, India.



feedstock

136

Secondary products

Nitrogen fertilisers

MAP/DAP

CAN/UAN/AS

Industrial applications

355

216

103

Current NH₃ (> 70 %) is primarily made via the *Haber-Bosch* process based on **fossil-fuel derived** H₂ (grey ammonia) which is produced via **Steam Natural Gas/Methane Reforming** (SMR). But green NH₃ may be on the horizon if the H₂ is made by other means.

NH



5

Key role of ammonia

- Ammonia is an essential global commodity.
- $\sim 85\%$ of ammonia is used for synthetic nitrogen fertilizer.

Hydrogen dependency

• Ammonia production accounts for 45% of global hydrogen consumption.

Key concepts about the role of NH_3

• Second-largest hydrogen user after the refining industry.

Decarbonisation potential

- Transition to "renewable ammonia" (i.e. from renewable H₂) supports chemical sector decarbonization.
- Promising applications: zero-carbon fuel (maritime sector, stationary power); H₂ carrier (long-range transport).











Ammonia can be an option for the Energy Transition

- Versatile hydrogen carrier → simplifies storage and transport; no need to return dehydrogenated carrier.
- **50 % cheaper** to produce from electricity than synthetic hydrocarbons or methanol
- Overall round-trip efficiency: 68 % (similar to LOHCs).
- Challenge as a fuel → difficult combustion; requires equipment modifications.









The LNG import and storage facilities (existing) can be converted to ammonia services but require additional (<u>and relevant</u>) design considerations (materials, structural design, insulation, safety devices).





Mature ammonia storage methods are currently available and the main factor determining the type of storage is the **ammonia storage capacity** and **economics**.

Туре	Pressure range, bar	Design temperature, °C	Capacity, t NH_3	Refrigeration compressor
Pressure storage	7 ÷ 18	Ambient temp.	< 1500	None
Semi-refrigerated storage	2 ÷ 4	- 9 ÷ 4	450 ÷ 2700	Single stage
Low-temperature storage	Close to 1 bar	- 33	4500 ÷ 45000	Two-stage
Solid-state storage	1 ÷ 30	20 ÷ 250	-	-



Ammonia terminals involve a combination of pressurized spheres (rail, road) and refrigerated storage tanks (ships).





There are **three main strategies** of NH₃ bunkering:

- 1. Truck to Ship
- 2. Ship to ship
- 3. Terminal to Ship

Each comes with safety concerns:

- 1. frequent connection/ disconnection operations;
- 2. spills on water;
- 3. large hold-ups.

Bunkering operations are potentially hazardous due to **high likelihood of leakage** of NH₃.

Criticalities:

- Risk assessment guidelines for ammonia bunkering.
- Need for specific risk assessment and safety barriers.
- Gas dispersion analysis and determination of controlled zones.



Ammonia fueled ships







- Colourless (but with a strong smell), reactive and lighter-than-air gas that dissolves readily in water, releasing heat.
- Only marginally flammable but does explode within certain vapor concentration limits (16-25 %) and in the presence of strong ignition sources.
- **Highly toxic**, causing serious health effects at low exposure levels (IDLH = 300 ppm; AEGL-1 = 30 ppm; AEGL-2 = 220 ppm; AEGL-3 = 2700 ppm at 10 min) (NIOSH)
- Solutions of ammonia are **alkali** and **corrosive** when concentrated \rightarrow material degradation.
- Environmental hazard, aquatic life is harmed by ammonia (and by the water used to depress an ammonia cloud).

Ammonia releases can lead to soil contamination.







Density relative to air: similar to methane (~ 0.5-0.6)

Accidental releases: liquid entrainment and low temperatures create heavy aerosol clouds

Burning velocity: < 0.07 m/s (vs. > 0.3 m/s of hydrocarbons)

Minimum ignition energy: very high (> 600 mJ vs 0.1-0.2 mJ of hydrocarbons)

Autoignition temperature: higher than most heavier hydrocarbons

Difficult to combust in air, but can explode in confined spaces (> 16% in air) Critical scenarios: boiling liquid pools, flash fires Extinguishing agents: CO₂ or powders

Toxicity and ecotoxicity: several orders of magnitude higher than other energy carriers.



Toxicity-related events are critical in the case of ammonia



Università degli Studi DI PADOVA

Ammonia-related accidents (particularly in industrial settings) have been a significant concern due to their potential health and environmental impacts.



> 250 accidents (1970-2024) (ARIA database), ~ 100 since 2018 (eSPIRS)

> 100+ accidents in Europe (1986-2022) (eMARS)

(mostly related to ammonia as a refrigeration fluid)



The human consequences may involve

workers and the general public !

> Anhydrous NH₃ releases

- > Involving **pipelines**, marine and road transport
- > Accidents with fatalities (15-30%), injuries (50-70%)

> 80%

> Common targets: workers



- Equipment / component failure
- Corrosion / damaged piping
- Maintenance / procedures

(OSHA, 2017)

CONSEQUENCES

- Ammonia leak
- Toxic cloud dispersion
- Environmental contamination
- (Flash fires), explosions

DIPARTIMENTO **KEYWORDS MAP - SCIENTIFIC ARTICLES ON AMMONIA SAFETY DI INGEGNERIA**



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nd equipment (especially in **y**.

- As part of the transition, ammonia fuel and associated new systems and equipment (especially in non-industrial contexts) will present **new technical and system complexity**.
 - New and modified technical and non-technical skills
 - "New" occupational health hazards

Most process hazards related to ammonia revolve around the control of storage and handling conditions to **prevent the loss of containment**

- Effective and reliable safeguards \rightarrow engineering and administrative controls
- Competency and preparedness









DI INGEGNERIA INDUSTRIALE AMMONIA IGNITION PROBABILITY (TOXIC MATERIAL, MARGINALLY FLAMMABLE)





The contact with water evolves heat (extinguishing agents: CO₂, powder).

Aerosol clouds are denser than air (> 4-8 % aerosol). Immediate reactions with air humidity.

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Lower **storage temperature** reduces the severity (REFRIGERATED < SEMI-REFRIGERATED < PRESSURIZED)



INDOOR OPERATIONS Control of ventilation is only partly effective at preventing toxic concentrations Ammonia intake protection (high-risk spaces integrity) Restricted areas



Construction material selection to prevent degradation mechanisms Control of stress **corrosion** cracking and corrosion under insulation / **Cold service** Auxiliary equipment to handle **expected** (PSVs, PRVs; inerting/purging during maintenance and startup/shutdown) and **unexpected** process releases (early leak detection, automatic isolation, drip trays, water mist systems, spilt ammonia disposal) Measurement and control of **ammonia emissions** via detectors





NH₃ as a **key enabler** of the energy transition

• Efficient hydrogen carrier with expanding applications (and emerging risks)

Safety remains the main challenge

• Toxicity, handling complexity, and environmental risks must be properly addressed

Infrastructure and operational risk management must evolve

• LNG facility adaptation, transport safety, and proper siting and zoning

Process safety culture and continuous risk assessment

- Frequent hazard reviews and audits
- Proper management of abnormal situations and overfilling scenarios
- Follow-through on safety measures
- Human factors and emergency preparedness are critical
 - Training, real-time decision-making under hazardous conditions, and robust safety protocols

🔖 The path forward

- Balancing decarbonization & safety is key to ammonia's success
- Strong safety culture and risk perception and communication





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THANK YOU FOR YOUR ATTENTION!

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