Explosion and fire in a chemical plant

- Chemical plant; upper-tier establishment,
- Production of fertilizers,
- Ammonia production unit, heat exchanger (HE)
- Substance involved: H2

## Heat exchanger

- Year of production:
   a) drum: from 1971
   b) chambers: 2011
- Subject to Technical Office's Inspection

#### Process gas

Average composition of process gas flowing into the HE:
<u>H<sub>2</sub> - 61 %</u>
<u>CO<sub>2</sub> - 22 %</u>
<u>H<sub>2</sub>O - 15 %</u>
CH<sub>4</sub> - 0,4 %, Ar - 0,3 %,
CO - 0,3 %, N<sub>2</sub> - 1 %

Composition of the process gas was not different from its usual content

#### Parameters of work

#### Before HE:

- Gas temperature 146.7 °C
- Gas pressure 3.07 MPa
- Water temperature (cooling agent) 38.5 °C

#### After HE:

- Gas temperature 39.7 °C
- Water temperature 115.7 °C









#### Phase I

- Rapid expansion of the process gas with hydrogen at a pressure of approximately 3.0 MPa
- The gas stream from the ruptured cover bursts asymmetrically down the structure towards the concrete substrate at level 00.
- Process gas with a majority of hydrogen, despite very low activation energy, does not ignite and does not explode due to the accompanying water vapor and carbon dioxide (typical extinguishing agents).
- The phase I period lasts about 10 seconds (according to the witnesess).

#### Phase II

- Spatial hydrogen explosion in the air (characteristic flash and noise).
- Wave of overpressure of the explosion makes a loss in the radius (150 m) – smashed glass and falling out of windows.
- The detonation wave (pressure-sound) hits the nearest adjacent objects at the epicenter of the explosion.
- Estimated time of rapid linear pressure drop from 3 MPa to near atmospheric pressure is approximately 20 seconds.

#### Phase III

- Free, limited, jet fire of the remains of the hydrogen in the instlation
- Addition of nitrogen to gas to extinguish the "burner" which causes a gradual decrease in the burning intensity and flame length.
- Estimated volume of process gas outflow 15 000 m3, including about 10 800 m3 of hydrogen.

#### The consequences

- 9 injuries, no fatalities,
- Damages on site: > 2 mln Euro,
- Production stopped for 2 weeks
- The amount of the process gas released: totally app. 9 MG including I MG of explosive and flammable (H2, CH4)

### Causes of the accident

- Two independent expert's opinions have been carried out (Technology University of Warsaw and Office of Technical Inspections)
- Direct: thinning /weakening of part of the heat exchanger's wall due to <u>corrosion</u> <u>and erosion</u> in the presence of humid/moist CO2 in the specific process conditions (in the chamber);

### Corrosion processes

- Electrochemical aspect of the corrosion (general reaction):  $Fe + CO_2 + H_2O \Rightarrow FeCO_3 + H_2$
- Partial pressure of CO2: 0,048 MPa
- Temperature: 80°C
- pH 4 acid character (typical value for condensed steam saturated with CO2)

Estimated corrosion speed: 5,5 mm/yr

#### Loss calculation

 reconstruction of damaged equipment and equipment - EUR 2.2 million

lost profit of the plant - EUR 2.6 million

 the value of lost production in relation to the plan - EUR 18 million

- Updating the existing procedures/instruction on technical supervisions/checks,
- Elaboration of trainings in order to discuss the causes, course and consequences of the accident with the Staff (also management)
- Introduction of the monitoring equipment (video cameras) within the unit

- Equipping the chemical rescue unit with the telemetric system and wireless duplex communication
- Analysis of the possibility of registration of the conversations on rescue channel
- Analysis and inclusion in the investment/action plan modernization of the DSC system (distributed control system) for the ammonia installtion

- Risk re-assessment of all the posts in the ammonium unit (H&S issue)
- Analysis of possible equipping control rooms with radiotelephones with laryngophone
- Carrying out of a complex elaboration specifing the ways of protection of control room aginst explosions

- Analysis of the possibility of changes in the design of HE's chambers
- Identitication of innovative methods of testing the technical equipment in the real operating conditions (and accepted by the authorities) allowing replacing hydraulic pressure tests





# Thank you for attention

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